A New Validated Stability-Indicating RP-HPLC Method for the Estimation of Pitavastatin in Tablet Dosage Forms
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ABSTRACT
An accurate and stability-indicating high performance liquid chromatographic method was developed for quantification of pitavastatin in its tablet dosage forms. Ideal separation of the drug was achieved on an Agilent Eclipse XDB C18 column (150 x 4.6 mm; 5µm) by eluting with a mobile phase consisting of a mixture of phosphate buffer (pH 3.4) and acetonitrile (65:35 v/v) at a flow rate of 0.9 mL/min. The drug in the eluates was monitored at 244 nm. Under optimized conditions, the retention time obtained for the drug was 3.905 min. The calibration plot was linear in the concentration range of 25-150 µg/mL of the drug. The validation of the method was done by following the ICH guidelines. The proposed method could be applied for determination of pitavastatin in its tablet dosage forms without any interference from normal excipients. The method thus, is suitable for routine quality control analysis of pitavastatin.

Key words: Pitavastatin, Estimation, Tablets, Stability-indicating, HPLC.

INTRODUCTION
Pitavastatin, [(3R,5S)-7-(2-cyclopropyl-4-(4-fluorophenyl) quinolin-3-yl) - 3, 5-dihydroxy 6(E)-heptenoic acid],[1] is a member of the medication class of statins. Pitavastatin is a novel, synthetic 3-hydroxy-3-methylglutaryl coenzyme-A (HMG-CoA) reductase inhibitor [2-3], approved for the treatment of hypercholesterolemia. HMG-CoA enzyme inhibitors inhibit the synthesis of mevalonate, a rate limiting step in synthesis of cholesterol, resulting in the lowering of cholesterol.[4] It is a competitive inhibitor and exerts its potent pharmacological actions by strongly binding to the active sites on HMG-CoA reductase. It has more potent lipid lowering action than other statins [5].
A literature survey revealed that some analytical methods have been reported for the determination of pitavastatin in pharmaceutical dosage forms using spectrophotometry [6], HPLC [7], HPTLC [8-9], UPLC [10] and LC/MS [11]. We have developed a new accurate and precise stability–indicating RP-HPLC method with short retention and run times for the determination of pitavastatin in bulk drug samples and in tablet dosage forms. The developed method has been duly validated as per ICH guideline.

**MATERIALS AND METHODS**

**A. Drugs, chemicals, and solvents**

The pure reference sample of pitavastatin was obtained from Aizant Drug Research Solutions Pvt Ltd., Hyderabad. The commercial tablet formulation of pitavastatin ‘Flovas’ (2 mg) manufactured by IPCA Laboratories Ltd., Mumbai, was purchased from the local market. Potassium dihydrogen orthophosphate, orthophosphoric acid, HPLC grade acetonitrile and HPLC grade methanol were purchased from Rankem Fine Chemicals Ltd., Mumbai. HPLC grade water was prepared by using Millipore Milli-Q system.

**B. Equipment and chromatographic conditions**

A Waters Alliance liquid chromatograph (model 2695) fitted with a diode array detector (model 2996) and running on Empower2 data handling system was employed in the study. An Agilent Eclipse XDB C\textsubscript{18} column (150 x 4.6 mm; 5µ) was used for analyzing the drug. All the chromatographic runs were carried out by using a mobile phase consisting of a mixture of phosphate buffer (pH 3.4) and acetonitrile (65:35 v/v) in isocratic mode at a flow rate of 0.9 ml/min. The injection volume of the samples was 10 µL. The detector wavelength was set at 244 nm. The chromatographic run time was set as 8.0 min. Under these optimized conditions, the retention time obtained for pitavastatin was 3.905 min.

**C. Preparation of the phosphate buffer**

The phosphate buffer was prepared by dissolving 1.36 gm of potassium dihyrogen phosphate in a beaker containing 1000 mL of water and the contents were sonicated. The pH of the solution was then adjusted to 3.4 with ortho phosphoric acid. It was then filtered through a 0.45µ membrane filter.
D. Preparation of the mobile phase
The optimized mobile phase consisted of a mixture of the above-mentioned phosphate buffer (pH 3.4) and acetonitrile in the ratio of 65:35 v/v.

E. Preparation of the diluent
Methanol was used as the diluent.

F. Preparation of the working standard solution of pitavastatin
10 mg of pitavastatin reference standard was accurately weighed and transferred into a 10 mL volumetric flask. To this, 7 mL of methanol was added, sonicated for 5 minutes and the volume was made up with a further quantity of methanol. This was used as the standard stock solution. The working standard solution was prepared by diluting 1.0 mL of the standard stock solution to 10 mL with the diluent in a volumetric flask.

G. Calibration plot
Solutions of pitavastatin at different concentration levels including the working standard concentration were prepared with the diluent. Twenty microlitres of each concentration was injected three times into the HPLC system (n=3). The response was read at 244 nm and the corresponding chromatograms were recorded. From these chromatograms, the mean peak areas at the different concentration levels were calculated and the linearity plot of mean peak areas over their concentrations was constructed.

H. Estimation of the drug from the tablet dosage forms
Ten ‘Flovas’ (2 mg) tablets were crushed and ground to a fine powder. Tablet powder equivalent to 10 mg of pitavastatin was transferred into a 10 mL volumetric flask. 8 mL of diluent was added and sonicated for 30 min. The volume was made up with the diluent and the contents were mixed well. This mixture was filtered through a 0.45µ membrane filter (discarding the first few mL of the filtrate). 1 mL of the filtrate was transferred into a 10 mL volumetric flask and made up to volume with diluent. This solution was then chromatographed six times. From the chromatograms obtained, the average drug content in the formulation was calculated.

Figure 3. Chromatogram of pitavastatin from its tablet dosage form

RESULTS AND DISCUSSION
During the method optimization studies, various combinations and proportions of the solvents and buffers were examined on an Agilent Eclipse XDB C18 column for efficient separation of pitavastatin. Using a mobile phase consisting of a mixture of phosphate buffer (pH 3.4) and acetonitrile in the ratio of 65:35 v/v, a good resolution and baseline separation of the drug peak was obtained. All the chromatographic conditions were optimized by valuating the column efficiency parameters like theoretical plates and tailing factor (Table 1). Under these optimized conditions, the retention time obtained for pitavastatin was 3.905 min (Figure 2) in a run time of 8.0 min. The method was then validated as per the ICH guideline. The proposed method was also found to be applicable for the analysis of pitavastatin in tablet formulations.
Table 1. Optimized chromatographic conditions

<table>
<thead>
<tr>
<th>Stationary Phase</th>
<th>Agilent Eclipse XDB C18 (100 x 4.6 mm, 5µm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mobile Phase</td>
<td>Phosphate buffer : Acetonitrile =65:35 v/v</td>
</tr>
<tr>
<td>Flow Rate</td>
<td>0.9 mL/min</td>
</tr>
<tr>
<td>Column Temperature</td>
<td>30°C</td>
</tr>
<tr>
<td>Injection Volume</td>
<td>10 µL</td>
</tr>
<tr>
<td>Detection Wavelength</td>
<td>244 nm</td>
</tr>
</tbody>
</table>

A. Specificity
A good analytical method should be able to measure the analytes accurately in the presence of probable interferences from its solvent as well as from the excipients of its formulation. Figure 2 shows good chromatographic baseline separation of pitavastatin from its working standard solution. Figure 3 demonstrates that no interfering peaks were observed at the retention time of pitavastatin arising due to the excipients of its tablet.

B. Linearity
The calibration curve (n=3) constructed for the drug was linear over the concentration range of 25 – 150 µg/mL. The regression of the plot was computed by least square regression method and is shown in Figure 4. The correlation coefficient is greater than 0.99 and the %RSD at each concentration studied was less than 2.

C. Accuracy and precision
The accuracy of the method was determined by recovery experiments. The recovery studies were carried out and the percent recovery with its standard deviation were calculated (Table 2). The high percentage of recovery indicates that the proposed method is quite accurate. The precision of the method was demonstrated by inter-day and intra-day variation studies. Six replicate injections of sample solutions were made and the percent RSD was calculated (Table 3).

Table 2. Accuracy data of the proposed method

<table>
<thead>
<tr>
<th>Analyte</th>
<th>Amount of the analyte taken (µg/mL)</th>
<th>Mean recovery of the analyte (µg/mL) ± SD</th>
<th>% Mean recovery ± SD</th>
</tr>
</thead>
<tbody>
<tr>
<td>Pitavastatin</td>
<td>50</td>
<td>50.26 ± 0.31</td>
<td>100.52 ± 0.62</td>
</tr>
<tr>
<td></td>
<td>100</td>
<td>100.14 ± 0.01</td>
<td>100.14 ± 0.01</td>
</tr>
<tr>
<td></td>
<td>150</td>
<td>151.65 ± 0.03</td>
<td>101.10 ± 0.02</td>
</tr>
</tbody>
</table>

Figure 4. Linearity plot for pitavastatin
### Table 3. Precision data for the proposed method

<table>
<thead>
<tr>
<th>S.No.</th>
<th>Intra-day precision</th>
<th>Inter-day precision</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>3722681</td>
<td>3721986</td>
</tr>
<tr>
<td>2</td>
<td>3722486</td>
<td>3722045</td>
</tr>
<tr>
<td>3</td>
<td>3722687</td>
<td>3718926</td>
</tr>
<tr>
<td>4</td>
<td>3722415</td>
<td>3722694</td>
</tr>
<tr>
<td>5</td>
<td>3722105</td>
<td>3722468</td>
</tr>
<tr>
<td>6</td>
<td>3722098</td>
<td>3722195</td>
</tr>
<tr>
<td>Average</td>
<td>3722412</td>
<td>3721719</td>
</tr>
<tr>
<td>SD</td>
<td>263.18</td>
<td>1394.03</td>
</tr>
<tr>
<td>%RSD</td>
<td>0.007</td>
<td>0.037</td>
</tr>
</tbody>
</table>

### D. System suitability parameters

System suitability parameters were studied with six replicate injections of the standard solution and the results are presented in Table 4.

### Table 4. System suitability parameters of the proposed method

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Retention time (min)</td>
<td>3.905</td>
</tr>
<tr>
<td>Tailing factor</td>
<td>1.1</td>
</tr>
<tr>
<td>Theoretical plates</td>
<td>7955</td>
</tr>
<tr>
<td>HETP</td>
<td>0.0126</td>
</tr>
</tbody>
</table>

### E. Degradation studies

**Peroxide degradation:** 1 mL of stock solution of pitavastatin was transferred into a 10 mL volumetric flask. To that 1 mL of 20% hydrogen peroxide (H₂O₂) was added. The solution was kept for 30 min at 60°C. The resultant solution was diluted with diluent to obtain 100 µg/mL solution of pitavastatin. 10 µL of this solution was injected into the system and the chromatogram was recorded.

**Figure 5. Chromatogram of pitavastatin subjected to peroxide degradation**

**Acid - degradation:** 1 mL of stock solution of pitavastatin was transferred into a 10 mL volumetric flask. To it 1 mL of 2M hydrochloric acid was added and refluxed for 30mins at 60°C. The resultant solution was diluted with diluent to obtain 100 µg/mL solution of pitavastatin
pitavastatin. 10 µL of this solution was injected into the system and the chromatogram was recorded.

**Figure 6. Chromatogram of pitavastatin subjected to acid - degradation**

*Base - degradation:* 1 mL of stock solution of pitavastatin was transferred into a 10 mL volumetric flask. To it, 1 mL of 2 M sodium hydroxide was added and refluxed for 30 min at 60°C. The resultant solution was diluted with diluent to obtain 100 µg/mL solution of pitavastatin. 10 µL of this solution was injected into the system and the chromatogram was recorded.

**Figure 7. Chromatogram of pitavastatin subjected to base - degradation**

*Dry heat degradation:* The working standard solution of pitavastatin was placed in oven at 105°C for six hours. The resultant solution was diluted to 100 µg/mL solution and 10 µL was injected into the system and the chromatogram was recorded.

**Figure 8. Chromatogram of pitavastatin subjected to dry heat - degradation**
**Photo - degradation:** The photochemical stability of the drug was also studied by exposing the working standard solution of pitavastatin to UV light by keeping the beaker in UV Chamber for 7 days. The resultant solution was diluted to obtain 100 μg/mL solution of pitavastatin. 10 μL of this solution was injected into the system and the chromatogram was recorded.

![Figure 9. Chromatogram of pitavastatin subjected to photo - degradation](image)

**Neutral - degradation:** Stress testing under neutral conditions was studied by refluxing the drug in water for 6 hours at a temperature of 60°C. The resultant solution was diluted to get 100 μg/mL solution of pitavastatin and 10 μL was injected into the system and the chromatogram was recorded.

![Figure 10. Chromatogram of pitavastatin subjected to neutral - degradation](image)

**F. Method suitability**
The commercial tablet formulation, Flovas (2 mg) was analyzed by the proposed method. The recovery obtained (100.1%) by the proposed method was found to be in good agreement with the labelled amount of the drug, which confirms the suitability of the method for the analysis of pitavastatin in tablet dosage forms.

**CONCLUSION**
The proposed RP-HPLC method is sensitive, precise, accurate and stability indicating and can be used for the routine determination of pitavastatin in its tablet dosage forms.
ACKNOWLEDGEMENT
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REFERENCES

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